

TWO NEW DIHYDROCHALCONES FROM *LINDERA ERYTHROCARPA*

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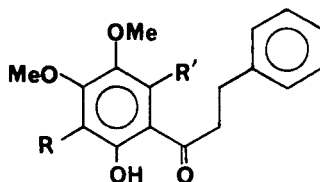
ABSTRACT.—Two new dihydrochalcones, dihydropashanone [**1**] and dihydrokanakugiol [**2**], were isolated from *Lindera erythrocarpa*. Their structures were elucidated by chemical and spectroscopic means.

The small deciduous tree, *Lindera erythrocarpa* (Lauraceae) Makino (Japanese name, "Kanakugi-no-ki"), is widely distributed in Japan, and its dried fruits are used as a folk medicine (digestive and anodyne drugs) (1). From the dried fruits of this plant, Liu *et al.* (2) isolated four pent-4-ene-1,3-dione derivatives, linderone, methyllinderone, lucidone, and methyllucidone, and later, Liu and Ogihara (1) isolated two flavonoids, kanakugiol and kanakugin.

We have reinvestigated the leaves, wood, and roots of *L. erythrocarpa*, and seven compounds were isolated. Five of the compounds were identified as 5,6-dehydrokawain (3), pinostrobin (3), methyl cinnamate, helilandin B (4,5), and pashanone (5,6) by direct comparison with the authentic samples. In this paper, we wish to describe the isolation and the structure elucidation of the remaining two compounds **1** and **2**.

Compound **1**, colorless prisms, mp 130–132°, gave a brown color with ethanolic FeCl₃, and the molecular formula was established by hrms (*m/z* 302.1160; C₁₇H₁₈O₅). The ir spectrum (3475 and 1635 cm⁻¹) indicated the presence of at least one phenolic hydroxyl group and a conjugated ketone group. The ¹H-nmr spectrum displayed an ethylene group at δ 3.03 and 3.42, two methoxyl groups at δ 3.83 and 3.88, an aromatic proton at δ 6.06 (1H, s), a hydroxyl group at δ 6.78, a phenyl group at δ 7.19–7.31, and a chelated hydroxyl group at δ 13.42. These data suggested that this compound is a 2'-hydroxydihydrochalcone with two methoxyl groups and two hydroxyl groups on the A ring, and it was suspected to be a dihydro derivative of one of the chalcones known in this plant. As pashanone was the only isomer isolated from this plant having two methoxyl groups and two hydroxyl groups on the A ring, we thought the compound may be a dihydro derivative of pashanone. Catalytic hydrogenation of pashanone gave 2',6'-dihydroxy-4',5'-dimethoxydihydrochalcone [**1**], which was indistinguishable from the natural product in all respects. This compound is known as dihydropashanone (7), but this is the first time it has been reported from a natural source.

Compound **2**, a viscous oil, gave a brown color with ethanolic FeCl₃, and the molecular formula was established by hrms (*m/z* 346.1418; C₁₉H₂₂O₆). The ir spectrum (1620 cm⁻¹) indicated the presence of a conjugated ketone group. The ¹H-nmr spectrum displayed an ethylene group at δ 3.03 and 3.38, four methoxyl groups at δ



- 1** R=H, R'=OH
2 R=R'=OMe

3.79, 3.86, 3.91, and 4.08, a phenyl group at δ 7.18–7.33, and a chelated hydroxyl group at δ 13.03. These data suggested that this compound is a 2'-hydroxy-3',4',5',6'-tetramethoxydihydrochalcone. Kanakugiol was hydrogenated over Pd-C to afford the dihydrochalcone, and we named this new compound dihydrokanakugiol [2].

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—All mps are uncorrected. Cc was run on Merck Si gel 60 (70–230 mesh) and florisil (100–200 mesh). Tlc was performed on glass plates precoated with Kieselgel 60 F₂₅₄ (Merck). Ms were recorded on a Hitachi M-52 spectrometer and hrms on a Hitachi M-80 spectrometer. Ir spectra were obtained on a JASCO IR-810 spectrophotometer and uv spectra on a JASCO UVIDEC-410 double-beam spectrophotometer. ¹H-nmr spectra were recorded on a JEOL JNM-GX-270 spectrometer operating at 270 MHz and ¹³C-nmr spectra on a JEOL JNM-FX-100 Fourier transform spectrometer at 25 MHz, with TMS as an internal standard. Chemical shifts are quoted in ppm.

EXTRACTION AND ISOLATION.—*L. erythrocarpa* was collected at Motosu-gun, Gifu prefecture, Japan (where a voucher specimen is deposited at the University), in July 1987. The plant material was divided into three parts, leaves (3.9 kg), wood (13.3 kg), and roots (6.8 kg), and then extracted with MeOH.

The MeOH extract of leaves was divided into hexane- and CHCl₃-soluble fractions. The former fraction (30.0 g) was chromatographed on a column of florisil. Elution with C₆H₆ gave dihydropashanone (35 mg), methyl cinnamate (12 mg), helilandin B (10 mg), and dihydrokanakugiol [2] (24 mg). Elution with C₆H₆-EtOAc (10:1) gave kanakugiol (367 mg), methylindrone (144 mg), methyllicudone (318 mg), and kanakugin (25 mg). The latter fraction (30.0 g) was chromatographed on a column of florisil. Elution with C₆H₆ gave methyllicudone (878 mg), dihydropashanone [1] (75 mg), pashanone (11 mg), and pinostrobin (50 mg).

The MeOH extract of wood was divided into hexane- and CHCl₃-soluble fractions. The former fraction (41.2 g) was chromatographed on a column of florisil. Elution with C₆H₆ gave kanakugiol (1.632 g), methylindrone (513 mg), methyllicudone (61 mg), and kanakugin (14 mg). The latter fraction (36.4 g) was chromatographed on a column of SiO₂. Elution with CHCl₃ gave kanakugiol (504 mg), methylindrone (134 mg), methyllicudone (61 mg), kanakugin (135 mg), dihydropashanone [1] (5 mg), and pashanone (20 mg).

The MeOH extract of roots was added to hexane, and the resulting precipitate was collected by filtration. The solid was recrystallized from MeOH to give pinostrobin (744 mg), and the hexane solution was concentrated to give a viscous oil (15.2 g). The residual MeOH solution was extracted with CHCl₃, and the CHCl₃ extract was concentrated to give a viscous oil (23.0 g). The hexane-soluble fraction was chromatographed on a column of florisil. Elution with C₆H₆ gave pinostrobin (1.698 g). Elution with C₆H₆-EtOAc (10:1) gave kanakugiol (279 mg), methylindrone (5 mg), pinostrobin (3.194 g), and 5,6-dehydrokawain (462 mg). The CHCl₃-soluble fraction was chromatographed on a column of SiO₂. Elution with CHCl₃ gave kanakugiol (48 mg), methylindrone (67 mg), kanakugin (44 mg), pinostrobin (375 mg), and 5,6-dehydrokawain (3.154 g).

The identification of helilandin B and pashanone was made by direct comparison with synthetic samples (5), and methylindrone, methyllicudone, kanakugiol, and kanakugin were identified by comparison of the spectral data and physical data with those published in the literature (1,8).

DIHYDROPASHANONE [1].—Colorless prisms; mp 130–132° (from CHCl₃); eims *m/z* [M]⁺ 302, 287, 197, 105; hrms *m/z* 302.1160 (calcd for C₁₇H₁₈O₅, 302.1153); ir ν max (CHCl₃) 3475, 3020, 1635, 1600, 1510 cm⁻¹; uv λ max (EtOH) 212, 233 (sh), 284, 341 nm; uv λ max (EtOH + 5% NaOH) 210, 246 (sh), 294, 375 nm; ¹H nmr (CDCl₃) δ 3.03 (2H, t, *J* = 7.7 Hz, β -H), 3.42 (2H, t, *J* = 7.7 Hz, α -H), 3.83, 3.88 (6H, s \times 2, 2 \times OMe), 6.06 (1H, s, 3'-H), 6.78 (1H, s, 6'-OH), 7.19–7.31 (5H, m, Ar-H), 13.42 (1H, s, 2'-OH); ¹³C nmr (CDCl₃) δ 204.5 (C=O), 162.0 (C-4*), 158.1 (C-6*), 151.3 (C-2*), 143.5 (C-1⁺), 141.6 (C-3⁺), 128.5 (C-2,3,5, and 6), 126.0 (C-4), 103.8 (C-1'), 92.7 (C-5'), 61.4 (OMe), 56.0 (OMe), 45.3 (C- α), 30.5 (C- β). (Values with an * or a ⁺ are interchangeable.)

DIHYDROKANAKUGIOL [2].—Viscous oil; eims *m/z* [M]⁺ 346, 331, 241; hrms *m/z* 346.1418 (calcd for C₁₉H₂₂O₆, 346.1415); ir ν max (CHCl₃) 3020, 1620, 1605 cm⁻¹; uv λ max (EtOH) 214, 282, 347 nm; ¹H nmr (CDCl₃) δ 3.03 (2H, t, *J* = 7.4 Hz, β -H), 3.38 (2H, t, *J* = 7.4 Hz, α -H), 3.79, 3.86, 3.91, 4.08 (12H, s \times 4, 4 \times OMe), 7.18–7.33 (5H, m, Ar-H), 13.03 (1H, s, OH); ¹³C nmr (CDCl₃) δ 205.6 (C=O), 154.4 (C-4*), 153.6 (C-6*), 151.2 (C-2*), 141.4 (C-1), 138.1 (C-5⁺), 136.9 (C-3⁺), 128.5 (C-2,3,5, and 6), 126.0 (C-4), 110.4 (C-1'), 61.3 (2 \times OMe), 61.0 (2 \times OMe), 45.2 (C- α), 30.4 (C- β). (Values with an * or a ⁺ are interchangeable.)

CATALYTIC REDUCTION OF PASHANONE.—A suspension of pashanone (3 mg) and 10% Pd-C (3 mg) in *i*PrOH (1 ml) was stirred under an H₂ atmosphere until uptake had ceased. The reaction mixture was filtered off and evaporated to dryness to give colorless prisms (3 mg). All the spectral data of this compound were indistinguishable from those of dihydropashanone [1].

CATALYTIC REDUCTION OF KANAKUGIOL.—A suspension of kanakugiol (22 mg) and 10% Pd-C (11 mg) in MeOH (3 ml) was stirred under an H₂ atmosphere until uptake had ceased. The reaction mixture was filtered off and evaporated to dryness to give a colorless oil (11 mg). All the spectral data of this compound were indistinguishable from those of dihydrokanakugiol [2].

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Received 1 March 1988